

Investigation of the impact of threshold pressure gradient on gas production from hydrate deposits

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Abstract

Many hydrate-bearing sediments are featured with unconsolidated argillaceous siltstones, which exist the non-Newtonian flow and threshold pressure gradient due to the high content of clay. In this study, the threshold pressure gradient of hydrate reservoir in the South China Sea is experimentally clarified. The quantitative relationship between it and the reservoir parameters is established. The function of threshold pressure gradient has been added into TOUGH+HYDRATE simulator. Here, site SH2, a candidate for field testing comprising a clayey silt gas hydrate reservoir in the Shenhu area of the South China Sea, was chosen to investigate the effect of threshold pressure gradient on gas production behavior through numerical simulations. The simulation results show that threshold pressure gradient has a significant impact on gas extraction. When the experimental value was applied, the gas production was enhanced unexpectedly, where cumulative gas output doubled in 5 years. The active hydrate dissociating area (gas, water and hydrate coexist) has notably extended, accompanying with the expanding cool zone. Water blockage near well was relieved.

29 However, with the increasing of threshold pressure gradient, propagation of pressure
30 would be restrained seriously. The pressure and hydrate of far-field formations stay
31 “frozen”. The bottom water invasion was weakened.

32 **Keywords:** natural gas hydrate; threshold pressure gradient; argillaceous siltstones;
33 numerical simulation

34

35 **1. Introduction**

36 Gas hydrate are solid crystalline compounds, in which small gas molecules are
37 lodged within the lattices of ice-like crystals. Natural gas hydrate deposits involve
38 mainly CH₄, and vastly exists in the permafrost and subsea or deep ocean sediments
39 [1]. Estimates of global hydrate reserves are very high and about 2×10^{12} tons of oil
40 equivalent, which is equivalent to twice that of all other fossil fuels combined [2].

41 Gas production from natural gas hydrate reservoirs can be initialized by
42 changing the reservoir condition outside of the hydrate stability zone; the production
43 strategies include hydrate dissociation through thermal stimulation [3],
44 depressurization [4-6], chemical potential shift [7-8], and the combinations thereof [9].
45 Among these, depressurization and thermal stimulation are the most known
46 approaches to extract gas from natural hydrate zones [10]. Recently, some enhanced
47 gas recovery strategies have been proposed to improve gas production from hydrate
48 formation [11], for example the CO₂/N₂ mixture has been injected in the field of
49 North Slope of Alaska to obtain a higher methane recovery [12].

50 More recently, a gas production pilot testing had been successfully conducted in
51 Shenhu area of the South China Sea operated by China Geological Survey, in which
52 over 30×10^4 sm³ natural gas had been extracted during 60 days, with methane
53 content up to 99.5% [13]. Furthermore, the second production test was progressed
54 successively in March 2020 and achieved continuous gas production for 30 days, with
55 a total gas output of 86.14×10^4 sm³ [14]. According to the progress of natural gas
56 hydrate exploration and test in South China Sea, there are still a lot of challenges and

57 uncertainties for a controllable and commercial gas extraction, including the high cost
58 of offshore test and the complex multi-phase flow mechanism [15].

59 The studies of sedimentary features demonstrated that hydrate reservoirs in
60 Shenhu area of the South China Sea are typical disperse type of hydrate accumulation,
61 where hydrate is distributed in unconsolidated argillaceous siltstone (or fine-grained)
62 sediments [16-20]. The sediment grain size analysis indicated that the content of
63 coarse-grained sand was less than 10% and clayed silt was up to 90% [21]. The
64 research suggests that the flow in sediments with high clay content usually exhibits
65 non-Darcy behavior [22]. Clay particles in sediments are flaky and hydrophilic, with
66 large specific area and high surface potential [23]. With the influence of micro-electric
67 field, the bound water strongly attracted to the surface of clay particles has a solid-like
68 property, which can cause great resistance to the water flow [24]. It is shown that, the
69 pore water flow in fine-grained formations with high clay content departs further from
70 the characterization of Navier-Stokes equation and appears as non-Newtonian
71 phenomena [25]. It cannot conform to Darcy's law. The relationship of flow velocity
72 and pressure gradient are nonlinear and there exists threshold pressure gradient (TPG).
73 The existence of TPG in porous media with high clay content was proposed by Miller
74 and Low [26]. Wu et al. [27] measured the threshold pressure of water phase for the
75 high-salt argillaceous dolomite reservoir in Jiangnan Basin. Wang et al. [28] showed
76 that TPG is negatively correlated with permeability and positively correlated with
77 water saturation based on the experiments and the relational fitted expression. Ren et
78 al. [29] proposed the permeability is the main controlling reason of the TPG, and
79 obtained the power function relationship between the TPG and the core permeability.

80 The hydrate reservoirs in the Shenhu area of the South China Sea are also typical
81 reservoirs with high argillaceous content and low permeability. Liu et al. [30] found
82 non-Darcy flow and TPG in the water phase seepage experiment of hydrate
83 argillaceous siltstone core in the South China Sea. He shows that pore water flows
84 only when the pressure gradient is greater than the minimum threshold pressure

85 gradient. Thus, it is necessary to find out the relations of TPG and flow conditions in
86 the argillaceous silt hydrate reservoirs in the South China Sea and further apply it to
87 hydrate exploitation. It is generally known that TPG can notably delay the change of
88 fluid velocity during the exploitation of low-permeability reservoirs, which restricts
89 the pressure propagation and thus accelerate water invasion at the later stage of
90 exploitation [31]. Ning et al. analyzed the simulation results of Sulige tight gas field
91 and discovered that the cumulative gas production would reduce by 16.7% under the
92 influence of TPG [32].

93 At present, there have been many numerical simulation studies targeting at
94 hydrate reservoirs in the China South Sea [33-37]. With the geological data at the first
95 offshore gas hydrate production test site, Sun et al. [38] built a 2D numerical
96 simulation model and investigated the short- and long-term production behavior of the
97 hydrate reservoir in Shenhu area. Qin et al. [6] focused on the detailed response of the
98 temperature and pressure in the first hydrate test in South China Sea. Yang et al. [39]
99 emphasized the effect of the flow capability of overlying and underlying strata on
100 hydrate production behaviors in depressurization-induced production of three-phase
101 hydrate deposits using a horizontal well, based on the geological data at the second
102 offshore gas hydrate production test site. Most of the scholars use
103 TOUGH+HYDRATE simulator in their research. This code was developed by the
104 Lawrence Berkeley National Laboratory, it can model the non-isothermal hydration
105 reaction, phase behavior, flow of fluids and heat under conditions typical of natural
106 gas hydrate deposits in complex formations. The model can handle any combination
107 of the possible hydrate dissociation mechanisms (i.e., depressurization, thermal
108 stimulation, and inhibitor-induced effects) [40]. However, none of the studies and
109 simulators has discussed the TPG existing in low-permeability hydrate reservoirs in
110 the South China Sea.

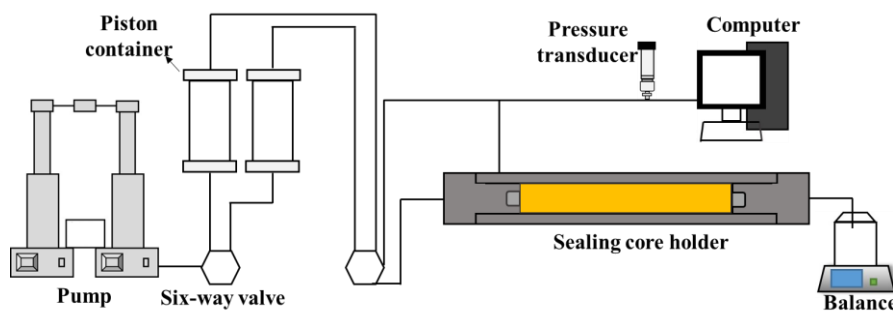
111 Therefore, in order to explore the characteristic of TPG in argillaceous silt
112 hydrate reservoirs in the South China Sea and its influence on the evolution of

113 temperature, pressure and fluid saturation during the exploitation, this study coupled
114 the multi-phase heat and mass transfer model, based on TOUGH+HYDRATE, with
115 the newly developed TPG mathematic model. The conservation equations have been
116 modified and the TPG module was added to TOUGH+HYDRATE, where the TPG
117 experimental result of target hydrate-bearing sediment in China South Sea was used.
118 A two-dimensional reservoir-scale production model based on the field data of SH2
119 drilling site was established for simulation. Meanwhile, the influence of TPG on the
120 spatial distribution of the main physical fields and the evolution of gas production
121 under depressurization were comparatively analyzed in detail.

122 2. Threshold pressure gradient model

123 2.1. Experiment setup

124 There exists a low velocity non-Darcy flow in the porous media with a high mud
125 content [40]. The threshold pressure gradient can characterize this phenomenon
126 concisely. In this study, the sediment sample obtained from pressure core of hydrate
127 reservoir in Shenhu area of the South China Sea was used in the flow experiment in
128 order to precisely quantify the TPG in argillaceous silt sediments.. Distilled water
129 combined with 3.5 wt% *NaCl* was used throughout the experiments to simulate
130 seawater. The experiment was carried out under room temperature. The schematic
131 diagram of the experimental apparatus for TPG measurement is shown in Fig. 1.
132 Through experimental fitting, the TPG of hydrate reservoir in the South China Sea is
133 clarified and the quantitative relationship between the TPG and the reservoir
134 parameters is established.

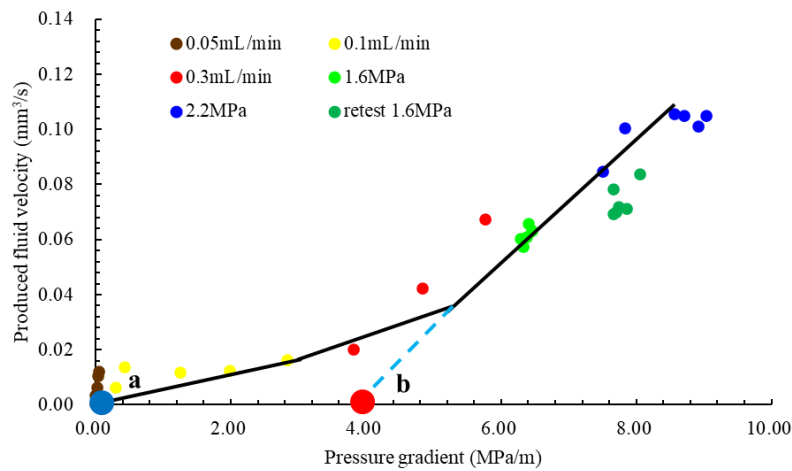


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136

Fig. 1 Schematic diagram of the threshold pressure gradient experiment process

137 This experimental method innovatively combines the improved micro flow
 138 method with the steady-state method. A group of cores can first measure the minimum
 139 threshold pressure gradient, and then continue to measure the quasi-threshold pressure
 140 gradient with the steady-state method. Thus, the minimum threshold pressure gradient
 141 and the quasi-threshold pressure gradient can be measured on one core at the same
 142 time, and the comparison and optimization of the two threshold pressure gradients are
 143 realized. By using the experimental method of combining the improved micro flow
 144 method with the steady-state method, a total of 26 groups of experimental tests of 13
 145 cores were completed.



146
 147 **Fig. 2 The relationship between the pressure gradient and the liquid production rate of the**
 148 **core BC08B-1-2**

149 Take the test data of core BC08b-1-2 as an example, we draw the relationship
 150 curve between pressure gradient and liquid production rate during core displacement,
 151 as shown in Fig. 2. According to the test principle of micro flow method, the
 152 corresponding pressure gradient at the moment before liquid outlet at the core outlet is
 153 the minimum threshold pressure gradient, that is, the pressure gradient corresponding
 154 to the starting point of pressure gradient and liquid production velocity curve is the
 155 minimum threshold pressure gradient. According to the test principle of steady-state
 156 method, on the relationship curve between pressure gradient and liquid production
 157 velocity, the straight-line section of the extended stable non-Darcy seepage curve

158 intersects with the coordinate axis, and the intersection point is the quasi-threshold
159 pressure gradient. Therefore, it can be seen from the analysis of Fig. 2 that the
160 minimum threshold pressure gradient of BC08b-1-2 core is 0.026 MPa/m (point **a** in
161 the figure corresponds to the pressure gradient), and the quasi-threshold pressure
162 gradient is 4.00 MPa/m (point **b** in the figure corresponds to the pressure gradient). In
163 our experiment, the produced fluid was measured by electron balance method, and the
164 accuracy is 0.01 g. Due to micro flow and steady-state displacement, the error of
165 minimum threshold pressure gradient and quasi-threshold pressure gradient depend on
166 the accuracy of the pressure sensor, which is 0.001 MPa for the measurement range of
167 0-5 MPa.

168 Using the above data analysis method, the minimum threshold pressure gradient
169 and quasi-threshold pressure gradient corresponding to 13 core displacement
170 experiments are obtained, as shown in Table 1. Take core BC08b-1-2 as an example,
171 steady-state data of 1.6 MPa and 2.2 MPa (small blue points and green points) are
172 used to calculate the permeability, and the average absolute deviation we evaluated
173 is 1.63%. Our experiments also showed that: 1) the higher the confining pressure
174 applied to the core, the lower the permeability K value obtained from the test, which
175 reflects that the higher the internal compaction degree of the core, the denser the core,
176 and the worse the seepage capacity; 2) The average value of the minimum threshold
177 pressure gradient of the core is 3.95 MPa/m, while the average value of the
178 quasi-threshold pressure of the core is 15.4 MPa/m, and the quasi-threshold pressure
179 gradient is about 4 times of the minimum threshold pressure gradient, which reflects
180 that the threshold pressure gradients obtained by the two test methods are quite
181 different. At present, the numerical range of the commonly used quasi-threshold
182 pressure gradient is high, and the minimum threshold pressure gradient reflects the
183 lower limit of the effective pressure gradient at both ends of the core when the fluid in
184 the core begins to flow. Compared with the quasi-threshold pressure gradient, its
185 physical significance and numerical range are more consistent with the actual

186 production situation. It is suggested to select reasonable threshold pressure gradient
 187 data according to the actual production situation.

188 **Table 1 The minimum threshold pressure gradient and quasi-threshold pressure gradient of**
 189 **different cores**

Core number	K (mD)	Minimum threshold pressure gradient (MPa/m)	Quasi-threshold pressure gradient (MPa/m)
BC08B-1-2	0.0195	0.026	4
BC08B-1-4	0.014	0.036	7.6
BC08B-1-8	0.00335	3.61	15
BC06B-1-2	0.0205	1.02	2.4
BC06B-1-4	0.0122	2.55	6
BC06B-1-8	0.00305	5.1	17
BC06B-1-12	0.0021	11.9	20
BC08B-2-8	0.0043	6.1	20
BC08B-2-12	0.0025	8.3	32
BC06B-2-2	0.00835	0.94	11
BC06B-2-4	0.0068	3.1	12
BC06B-2-8	0.0031	1.5	25
BC06B-2-12	0.0018	7.2	28
Mean value		3.95	15.4

190

191 The experimental results demonstrate that the relation of TPG and
 192 permeability-viscosity ratio presents a power function. Using power function
 193 regression analysis, the relationship equation between the TPG and
 194 permeability-viscosity ratio in hydrate reservoir of the South China Sea can be
 195 established as

$$196 \quad \lambda = 0.14 \left(\frac{k}{\mu} \right)^{-0.86} \quad (1)$$

197 where λ is the threshold pressure gradient, $\text{MPa}\cdot\text{m}^{-1}$; k is the permeability, $10^{-3}\mu\text{m}^2$; μ
 198 is viscosity of liquid, $\text{mPa}\cdot\text{s}$.

199 2.2 Selection of flow equation

200 According to the research results, there are mainly four forms of the
 201 mathematical models used to characterize the non-linear flow with TPG, as listed
 202 below.

203 $\vec{v} = a[\text{grad}P - \lambda]$ (2)

204 $\vec{v} = a[\text{grad}P - \lambda]^b$ (3)

205 $\vec{v} = a[\text{grad}P]^2 + b \cdot \text{grad}P + c$ (4)

206 $\vec{v} = a[\text{grad}P - b]$ (5)

207 where \vec{v} is fluid flow velocity, cm/s; $\text{grad}P$ is pressure gradient, 10^{-4} MPa/m; a , b , c
 208 are regression coefficients of equation.

209 Among them, Eq. (2) is well consonant with the experimental observation, and
 210 can better reflect the nonlinear flow characteristics with TPG. In the establishment of
 211 flow equation, Bingham plastic fluid simulation method has been used for calculation.
 212 We can still use Darcy's law to describe its flow process by using the effective
 213 potential gradient method [41], as shown in Eq. (6).

214
$$\vec{v} = -\frac{k}{\mu} \left[1 - \frac{\lambda}{|\text{grad}p|} \right] \cdot \text{grad}p \quad |\text{grad}p| > \lambda$$
 (6)

$$\vec{v} = 0 \quad |\text{grad}p| < \lambda$$

215 where k is the permeability, μm^2 ; μ is viscosity of liquid, mPa·s.

216 3. Simulator improvement

217 The motion equation coupled with threshold pressure gradient is the evolvement
 218 of classical Darcy's equation [42]. In particular, a series of research results based on it
 219 can effectively guide the exploitation of argillaceous silt hydrate reservoirs in the
 220 South China Sea. In order to establish a more reliable flow model for hydrate
 221 exploitation in argillaceous silt sediments, this part combined the previously
 222 developed multi-phase flow equation and new TPG experimental model.

223 3.1 Conservation equation modification

224 Based on the TOUGH+HYDRATRE code, the general formula for mass and heat
 225 balance considerations is described as [43]

$$226 \quad \frac{d}{dt} \int_{V_n} M^K dV = \int_{\tau_n} F^K n d\tau_n + \int_{V_n} q^K dV \quad (7)$$

227 where M^K is mass accumulation term of component; K are different components (i.e.,
 228 water, methane, and water-soluble inhibitors); F^K is flux vector of component; q^K is
 229 source/sink term of component.

230 The mass fluxes of water, CH₄ and inhibitor include contributions from the
 231 aqueous and gaseous phases, i.e.,

$$232 \quad F^K = \sum_{\beta=A, G} F_\beta^K \quad (8)$$

233 For the aqueous phase, the phase flux \mathbf{F}_A is described by Darcy's law as:

$$234 \quad F_A = - \frac{kk_{rA}\rho_A}{\mu_A} X_A^K (\nabla P_A - \lambda) \quad (9)$$

235 where k is absolute permeability; k_{rA} are relative permeability of liquid phase; μ_A are
 236 viscosity of the liquid phase; P_A is aqueous pressure; X_A^K is mass fraction of
 237 component K in liquid phase; λ is threshold pressure gradient.

238 Considering that the gaseous flow would not be affected by TPG, the relevant
 239 modification aimed to flux vector of component in liquid phase.

240 Among them, the mass conservation equation of water w can be given by

$$241 \quad \int \frac{\partial}{\partial t} (\phi S_A \rho_A X_A^w + \phi S_I \rho_I + \phi S_G \rho_G X_G^w + \phi S_H \rho_H X_H^w) = X_A^w q_A + X_G^w q_G \quad (10)$$

$$+ \nabla \cdot \left[\frac{kk_{rA}\rho_A X_A^w}{\mu_A} (\nabla P_A - \lambda) + \frac{kk_{rG}\rho_G X_G^w}{\mu_G} (\nabla P_G) \right]$$

242 where subscript A represents liquid phase; subscript I represents ice phase; subscript
 243 G represents gas phase; ϕ is porosity; S_α is saturation of phase α ; ρ_α is density of
 244 phase α ; X_α^w is mass fraction of component w in phase α ; q_A is mass of productive
 245 liquid phase; q_G is mass of productive gaseous phase; Q^w is mass of hydrate-related
 246 release of water; k is absolute permeability; k_{rA} , k_{rG} are relative permeability of
 247 liquid and gaseous phase; μ_A , μ_G are viscosity of the liquid and gaseous phase; P_A and
 248 P_G are aqueous pressure and partial methane pressure, respectively; λ is threshold
 249 pressure gradient.

250 The mass conservation equation of methane m in liquid phase can be defined as

$$\begin{aligned}
 & \int \frac{\partial}{\partial t} (\phi S_A \rho_A X_A^m + \phi S_G \rho_G X_G^m + \phi S_H \rho_H X_H^m) = X_G^m q_G + X_A^m q_A \\
 & + \nabla \cdot \left[\frac{k k_{rA} \rho_A X_A^m}{\mu_A} (\nabla P_A - \lambda) + \frac{k k_{rG} \rho_G X_G^m}{\mu_G} (\nabla P_G) \right]
 \end{aligned} \tag{11}$$

252 where X_α^m is mass fraction of component m in phase α ; Q^m is mass of hydrate-related
 253 release of methane.

254 The mass conservation equation of water-soluble inhibitor i in liquid phase can
 255 be expressed as

$$\int \frac{\partial}{\partial t} (\phi S_A \rho_A X_A^i) = \nabla \cdot \left[\frac{k k_{rA} \rho_A X_A^i}{\mu_A} (\nabla P_A - \lambda) \right] + X_A^i q_A \tag{12}$$

257 where X_A^i is mass fraction of water-soluble inhibitors i in liquid phase.

258 These components are partitioned among four possible phases: gas phase, liquid
 259 phase, ice phase and hydrate phase. The hydrate in the reservoir turns into liquid
 260 phase and gas phase at the phase transition condition, and ice might be formed at
 261 subzero temperatures due to the endothermic effect of hydrate dissociation.

262 Considering the threshold pressure gradient, the heat conservation equation can
 263 be given by

$$\begin{aligned}
 & \int \frac{\partial}{\partial t} [(1-\phi) \rho_R C_R T + \phi S_A \rho_A U_A + \phi S_G \rho_G U_G + \phi S_H \rho_H U_H] = Q_{diss} + \\
 & \nabla \cdot [(1-\phi) K_R + \phi (S_A K_A + S_G K_G + S_H K_H)] \nabla T + \\
 & \nabla \cdot \left[h_A \frac{k k_{rA} \rho_A}{\mu_A} (\nabla P_A - \lambda) + h_G \frac{k k_{rG} \rho_G}{\mu_G} (\nabla P_G) \right]
 \end{aligned} \tag{13}$$

265 where subscript R represents rock matrix; C_R is heat capacity of the dry rock; U_α is
 266 specific internal energy of phase α ; Q_{diss} is equilibrium dissociation heat; K_R is
 267 thermal conductivity of the rock matrix; K_α is thermal conductivity of phase α ; h_A , h_G
 268 is the specific enthalpy of liquid and gaseous phase; T is temperature.

269 3.2 Threshold pressure gradient module

270 It can be revealed from the flow experiment that the threshold pressure gradient
 271 increases with the permeability of clayed sediment decreasing. When the permeability

272 is relatively high, the TPG increases slowly as the permeability decreases. When the
273 permeability is relatively low, the TPG increases sharply as the permeability decreases.
274 The TPG and the permeability-viscosity ratio agree with the power function.
275 Therefore, the relationship between TPG and the permeability-viscosity ratio has been
276 characterized through the power function formula in the newly-added TPG module.

$$277 \quad \lambda = a \left(\frac{k}{\mu} \right)^b \quad (14)$$

278 where a , b are feature parameters, b is usually negative. Parameter a can control the
279 dimension of TPG and b can determine the relationship between the TPG and
280 reservoir parameters. Enter the values of a and b in this module and a specific TPG
281 curve can be given.

282 Experimental values ($a = 0.14$ and $b = -0.86$) were used in the subsequent
283 simulation, which was more suitable and fitted in with the actual circumstances.

284 **4. Model setup**

285 4.1 Geological background

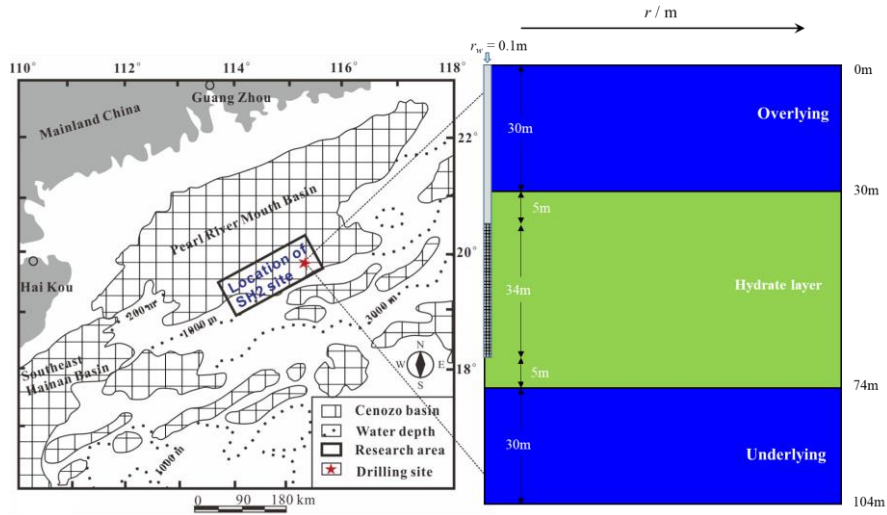
286 Shenhu area is in the middle of the northern slope of South China Sea and
287 tectonically located in the Zhu II Depression, Pearl River Mouth Basin [44, 45]. This
288 area has already become a large source of oil and natural gas. Numerical simulations
289 in this study are based on gas hydrate reservoir at the site SH2. In 2007, gas hydrate
290 samples were collected at the site SH2 during a scientific expedition conducted by the
291 Guangzhou Marine Geological Survey [46, 47]. The hydrate-bearing deposits at
292 drilling site SH2 are at a depth of 185-229 mbsf, with a porosity of 40% and an
293 average permeability of 10 mD. The seafloor is at a water depth of 1235 m, with a
294 temperature of 3.9°C. Hydrate saturation varies with depth. In situ measurements
295 indicated that the highest hydrate saturation reached 47% and the average hydrate
296 saturation was 16.5%. The gas produced from the hydrate consists of 96.1-99.82%
297 methane.

298 4.2 Model construction

299 In this study, a synthetic multi-phase flow models with TPG described above was
300 created with fundamental hydrate bearing sediments parameters collected from
301 Shenhu area in the South China Sea. In order to focus on TPG effect analysis and
302 minimize the influence from different aspects, the following assumptions have been
303 made in the numerical model: the pure methane is generated from hydrate; the
304 hydrate-bearing layers, overlying layers and underlying layers are homogenous, and
305 the mechanical properties of hydrate sediment stay unaffected during hydrate
306 dissociation process. In order to avoid the influence of grid size on the numerical
307 simulation results, prior to this study, we have performed grid refinement study to find
308 the desirable gride size. This information has now also been added in the revised
309 manuscript.

310 As shown in Fig. 3, the hydrate reservoir model consists of the three basic layers:
311 44 m thick methane hydrate layer, two 30 m thick overlying and underlying
312 formations with similar permeability conditions. It consists of 120 radial grid layers
313 and 110 vertical grid layers, with a total grid layer quantity of $120 \times 110 = 13200$. The
314 grids of the methane hydrate layers and the near-wellbore area are refined in order to
315 increase the accuracy of simulation.

316 The pressure and temperature at the bottom of the hydrate is calculated to be
317 15.22 MPa and 15.01 °C. The top and bottom pressure are respectively 14.5 MPa and
318 15.47 MPa. The top and bottom temperature are respectively 11.75 °C and 16.21 °C.
319 The overburden and underlying layer are fully saturated with water. The main
320 properties of the simulated hydrate bearing sediment are listed in Table 2 [48, 49].



321

322

323

Fig. 3 Schematic model of the methane hydrate production scenario

Table 2 Basic hydrate reservoir properties

Parameter	Value
Thickness of hydrate layer /m	44
Thickness of overburden layer /m	30
Thickness of underlying layer /m	30
Model radius /m	1000
Well radius /m	0.1
Permeability of overburden layer /mD	10
Permeability of hydrate layer /mD	10
Permeability of underlying layer /mD	10
Porosity of overburden layer	0.42
Porosity of hydrate and underlying layer	0.4
Initial hydrate saturation	0.165
Gas component	100%CH ₄
Grain density (kg/m ³)	2600
Compression coefficient (Pa ⁻¹)	1.0×10 ⁻⁸
Wet thermal conductivity (W/m/K)	3.1
Relative permeability mode	$k_{rG} = (S_G^*)^n$

$$k_{rA} = (S_A^*)^n$$

$$S_A^* = (S_A - S_{irA}) / (1 - S_{irA})$$

$$S_G^* = (S_G - S_{irG}) / (1 - S_{irA})$$

Capillary pressure model (van
Genuchten Model)

$$P_{cap} = -P_0 \left(\left[S^* \right]^{-1/\nu} - 1 \right)^{1-\nu}$$

$$S^* = (S_A - S_{irA}) / (S_{mxA} - S_{irA})$$

324 Where k_{rG} , k_{rA} are relative permeability of gaseous and liquid phase; S_G , S_A are saturation of
 325 gaseous and liquid phase; S_{irG} , S_{irA} are the irreducible gas and water; P_{cap} is capillary pressure; P_0
 326 = 10^5 Pa; $S_{mxA} = 1 - S_{irG}$ is maximum water saturation; ν is van Genuchten exponent.

327 4.3 Well design

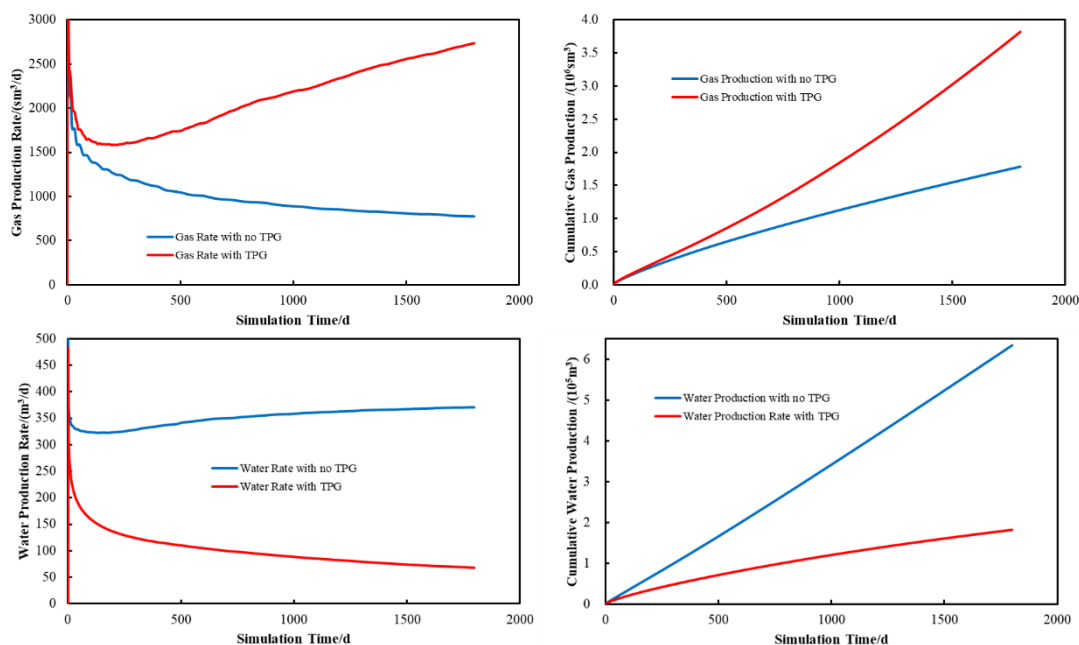
328 A vertical well with a diameter of 0.1 m was drilled for hydrate exploitation by
 329 depressurization. The perforation section is 34 m (-35 m to -69 m in the model). The
 330 production well is operated at a constant pressure of 4.5 MPa at bottom of the
 331 wellbore. The production period is 5 years.

332 5. Result and analysis

333 5.1 Production behavior

334 The simulation results of the cases applying TPG and no TPG are compared with
 335 each other in this part. Fig. 4 shows the comparison of gas and water production
 336 within 5 years. It can be seen from Fig. 4a that, when the TPG is not applied, gas
 337 production rate drops quickly from the peak value (> 3000 sm³/d) to 1266 sm³/d in the
 338 200th day and the decreases slowly to less than 800 sm³/d. When the TPG is applied,
 339 initial decline of gas production still exists but it only drops to 1580 sm³/d in the
 340 200th day. After that, the gas production rate begins to boom, which is absolutely
 341 different from the case without TPG. Of interest is the increase at late times and the
 342 gas production rate climbed to 2732 sm³/d at the end. Fig. 4b reveals that, when the
 343 TPG is not applied, the cumulative gas production in 5 years is only 178.4×10^4 sm³.
 344 While considering the TPG, the cumulative gas production reaches 382.1×10^4 sm³,

345 which is more than twice of the former. TPG also makes a distinct impact on water
 346 production. Fig. 4c demonstrates that, without TPG, the water production rate
 347 fluctuates at a stable and slowly increasing state in a long term. As production
 348 progresses, water saturation in the reservoir increases, resulting in increased water
 349 production rates in the later stage, which increases from 320 m³/d at initial to 370
 350 m³/d at the end. However, when TPG is applied, the water flow seems to be restrained.
 351 The water production reduces gradually to 68 m³/d and the cumulative water
 352 production is only 18.2 × 10⁴ m³, which is much lower than that without TPG (63.4 ×
 353 10⁴ m³).



354
 355 **Fig. 4 Performance of fluid productions applying TPG and no TPG**

356 Note that while a reduction in water production is expected because the threshold
 357 pressure can delay the pressure propagation, the evolution of gas output is contrary to
 358 expectation. It can be preliminarily concluded that TPG in argillaceous silt reservoir
 359 derived from flow experiment is propitious to enhance recovery. As quantified in Fig.
 360 5, the gas-water ratio with TPG gradually increases and can exceed 40 by the end of
 361 the simulation. In contrast, the gas-water ratio without TPG is close to 2. A too low
 362 gas-water ratio may cause difficulties to downhole gas-water separation, which is
 363 extremely unfavorable in practice.

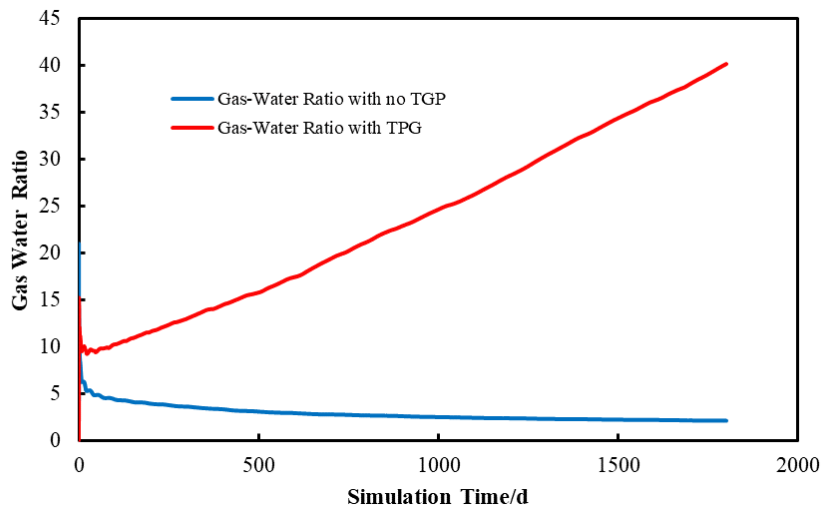
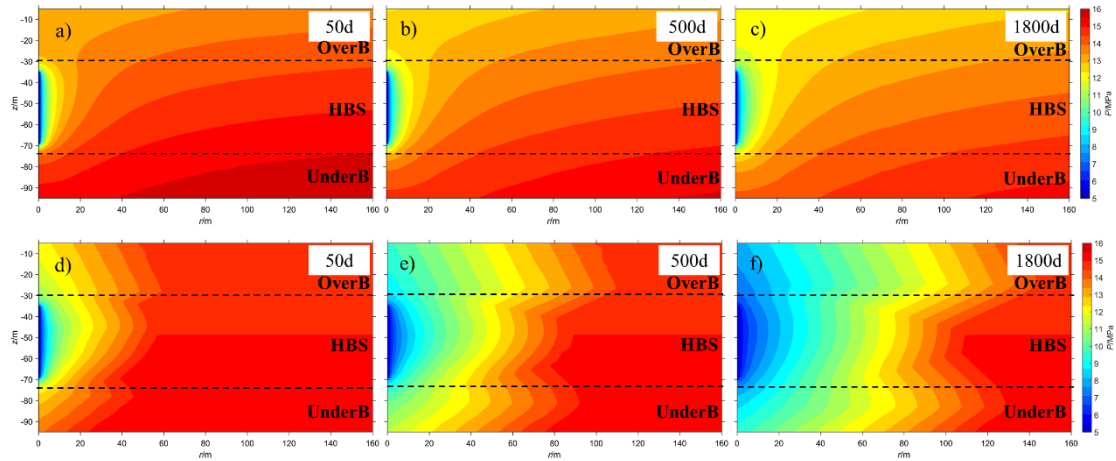


Fig. 5 Performance of Gas-water ratios applying TPG and no TPG

5.2 Evolution of reservoir parameters during production.

The reason for the influence of TPG on productivity can be further discussed by analyzing the continuous evolution of reservoir parameters during hydrate exploitation. The variations of the pressure, temperature and saturation during gas production are plotted in Fig. 6 to 10.

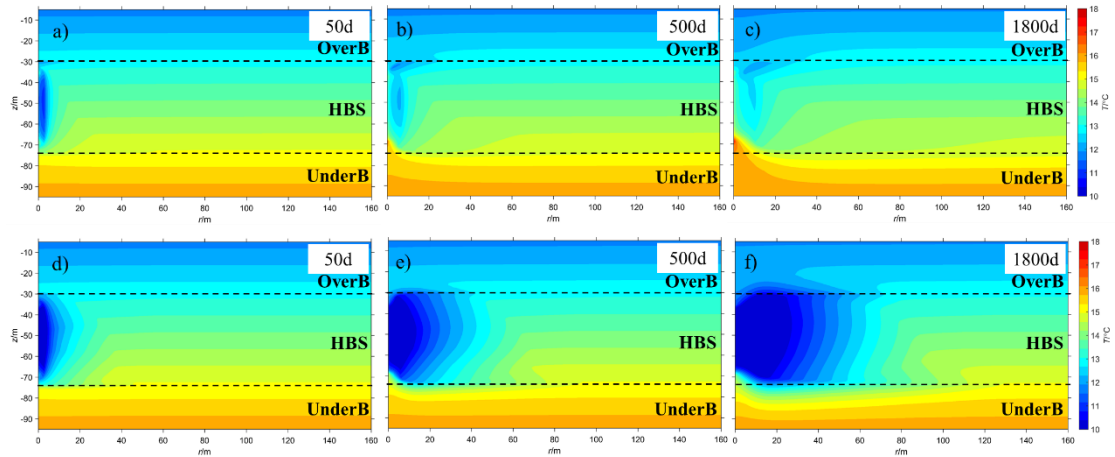
Review of the pressure profile in Fig. 6 shows the difference of pressure evolution caused by TPG. Fig. 6a, b and c reveal the distribution of pressure at the 50th, 500th, and 1800th day without TPG and Fig. 6d, e and f show those with TPG. Simulation results without TPG indicates that the pressure distribution conforms to expectations, as maximum pressure drops near the well and overall pressure decline in the reservoir over time. In Fig. 6c, pressure advance front looks approximately horizontal. Although the range of pressure drop is relatively larger, the degree of that is low. The front of pressure drop, which is larger than 10 MPa, advances less than 5 m. The most striking feature with TPG in Fig. 6d is the emergency of the radial extension of pressure advance front around the perforation section. While the pressure front reaches only 120 m, the pressures of underlying layer and overlying layer drop significantly and the front of the low-pressure area ($< 10\text{MPa}$) advances to 40 m, which is quite beneficial to the hydrate dissociation.



384

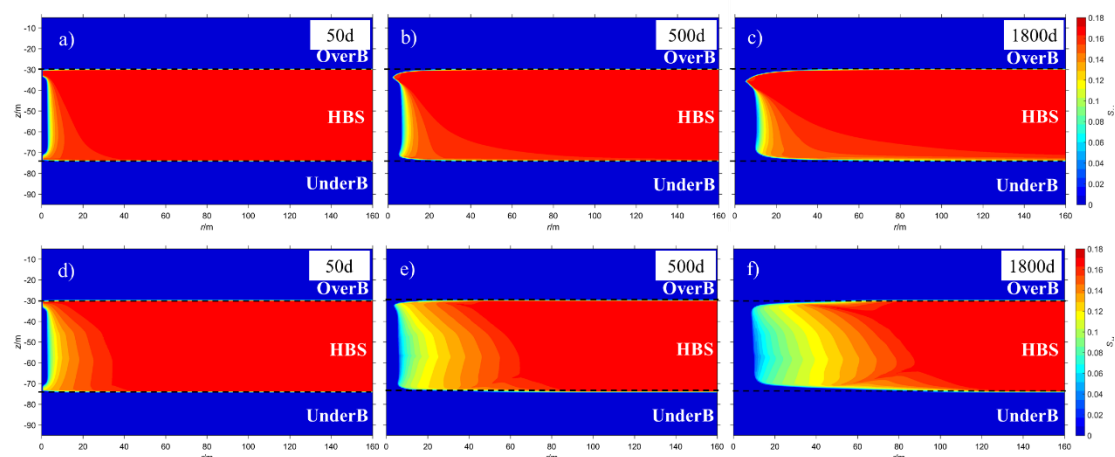
385 **Fig. 6 Effect of TPG on the pressure distribution (a-50d without TPG, b-500d without TPG,**
 386 **c- 1800d without TPG; d-50d with TPG, e-500d with TPG, f-1800d with TPG)**

387 Hydrate dissociation is a typical endothermic reaction. Review of the
 388 temperature profile in Fig. 7 can effectively reflect the dynamics of hydrate. Localized
 389 effects like the cooling in dissociating hydrate-bearing layers are easily recognizable.
 390 The expanding and sagging cool zones are caused by continuing dissociation,
 391 Joule-Thomson cooling, and drainage of the cooler water released from dissociation.
 392 Fig. 7a, b and c reveal the distribution of temperature at the 50th, 500th, and 1800th
 393 day without TPG and Fig. 7d, e and f show those with TPG. When TPG is not applied,
 394 the expanding colder zone adjacent to the overlying layer ($< 12^{\circ}\text{C}$) appears within 1-3
 395 m near the well in the early stage of production, as Fig. 7a confirm. Of interest is the
 396 shrinkage of the cool zone in the main hydrate body at 500d in Fig. 7b, when
 397 compared to that at 50d in Fig. 7a. It means that the amount of hydrate dissociating
 398 gradually decreases, which is consistent with the decline in gas production. Fig. 9c
 399 clearly shows that the bottom water intrusion from the underburned layer also brings
 400 about the temperature rise at the bottom of the production well. As depicted in Fig. 7d
 401 to 7f, considering TPG, the area with low temperature ($< 12^{\circ}\text{C}$) in the initial stage of
 402 production reaches 10 m and gradually expands as time advances. By the end of the
 403 simulation, the front of the low-temperature region has extended to 50 m, it is shown
 404 that the rise of bottom water affects the fall of temperature.



405
 406 **Fig. 7 Effect of TPG on the temperature distribution (a-50d without TPG, b-500d without**
 407 **TPG, c- 1800d without TPG; d-50d with TPG, e-500d with TPG, f-1800d with TPG)**

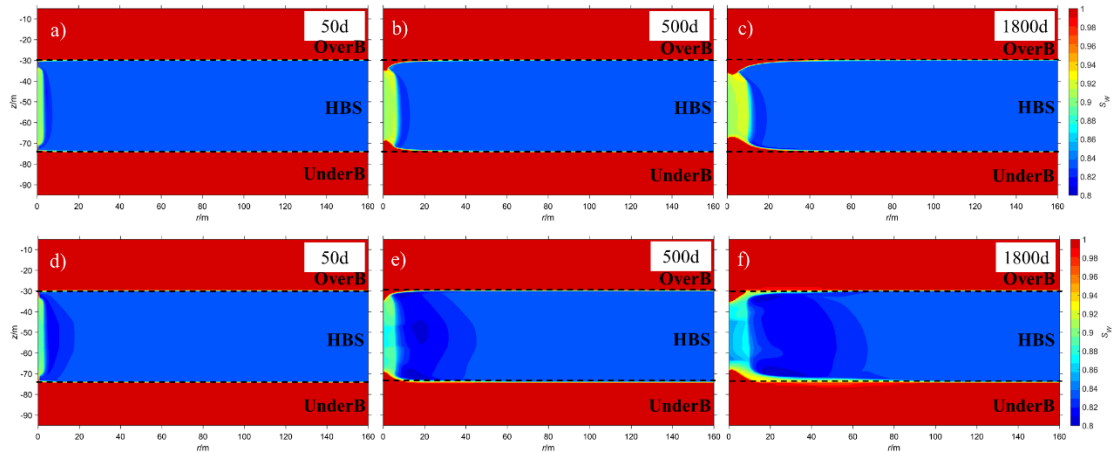
408 Some of the most interesting observations can be made from the distribution of
 409 the hydrate saturation in Fig. 8. Fig. 8a, b and c reveal the distribution of hydrate
 410 saturation at the 50th, 500th, and 1800th day without TPG and Fig. 8d, e and f show
 411 those with TPG. From the comparison between results without and with TPG, it can
 412 be seen that there is little difference in the range of the hydrate dissociated area where
 413 only gas and water phases exist. When TPG is applied, the advance speed of the front
 414 of the active dissociation regions, which reaches 80 m from well, is much higher than
 415 that without TPG. Fig. 8f shows clear signs of the markedly lower hydrate saturation
 416 in active dissociation regions.



417
 418 **Fig. 8 Effect of TPG on the hydrate distribution (a-50d without TPG, b-500d without TPG,**
 419 **c- 1800d without TPG; d-50d with TPG, e-500d with TPG, f-1800d with TPG)**

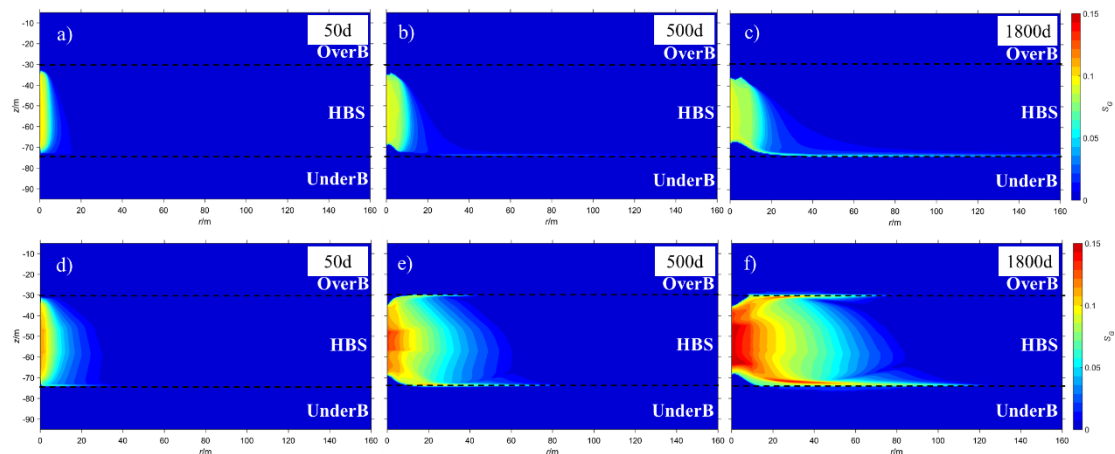
420 The TPG also has a great impact on the liquid and gaseous phase saturation. Fig.
 421 9a, b and c reveal the distribution of water saturation at the 50th, 500th, and 1800th

422 day without TPG and Fig. 9d, e and f show those with TPG. Fig. 9c shows clear signs
 423 of the zone with high water saturation around the production well, probably causing
 424 "water blockage". In contrast, Fig. 9f shows the emergence of a slightly
 425 water-saturated zone around the well. In Fig. 9f, Aquifers invading into hydrate layer
 426 are smaller in volume than that in Fig. 9c. Bottom water invasion may cause the
 427 increase of water production rate in the late time of simulation without TPG.



428
 429 **Fig. 9 Effect of TPG on the aqueous phase distribution (a-50d without TPG, b-500d without**
 430 **TPG, c-1800d without TPG; d-50d with TPG, e-500d with TPG, f-1800d with TPG)**

431 The evolution of gaseous phase saturation distributions in the reservoir are
 432 shown in Fig. 10. Fig. 10a, b and c reveal the distribution of gaseous phase saturation
 433 at the 50th, 500th, and 1800th day without TPG and Fig. 10d, e and f show those with
 434 TPG. When TPG is not applied, the gas saturation conforms to expectations, with
 435 maximum enrichment near well. When TPG is applied, the gas enrichment near well
 436 is enhanced.



437

438 **Fig. 10 Effect of TPG on the gas phase distribution (a-50d without TPG, b-500d without**
 439 **TPG, c-1800d without TPG; d-50d with TPG, e-500d with TPG, f-1800d with TPG)**
 440

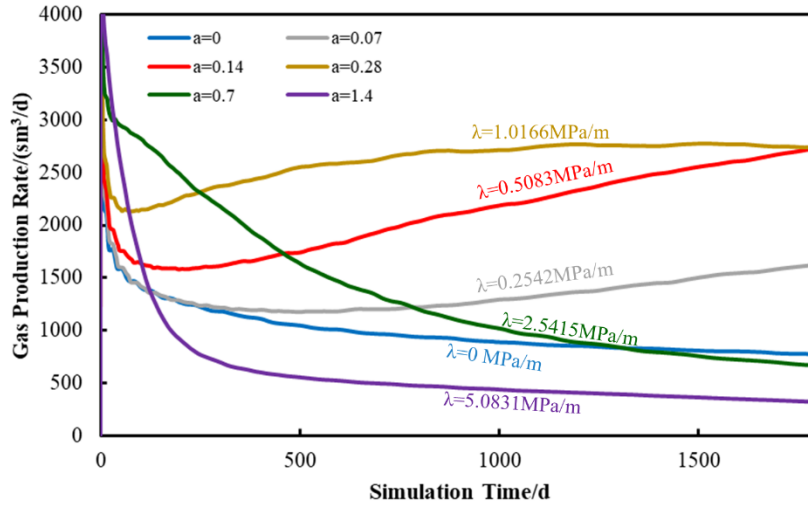
441 5.3 Influence of different TPG on evolution of system conditions

442 In order to evaluate the influence of different TPGs on the evolution of system
 443 conditions during production, the feature parameters in Eq. (14) were set with
 444 different values. Table 3 shows the corresponding TPGs with different a .

445 **Table 3 Model parameters and the corresponding TPG**

No.	Value of a	Corresponding TPG (MPa·m ⁻¹)
case 1	0	0
case 2	0.07	0.2542
case 3	0.14	0.5083
case 4	0.28	1.0166
case 5	0.7	2.5415
case 6	1.4	5.0831

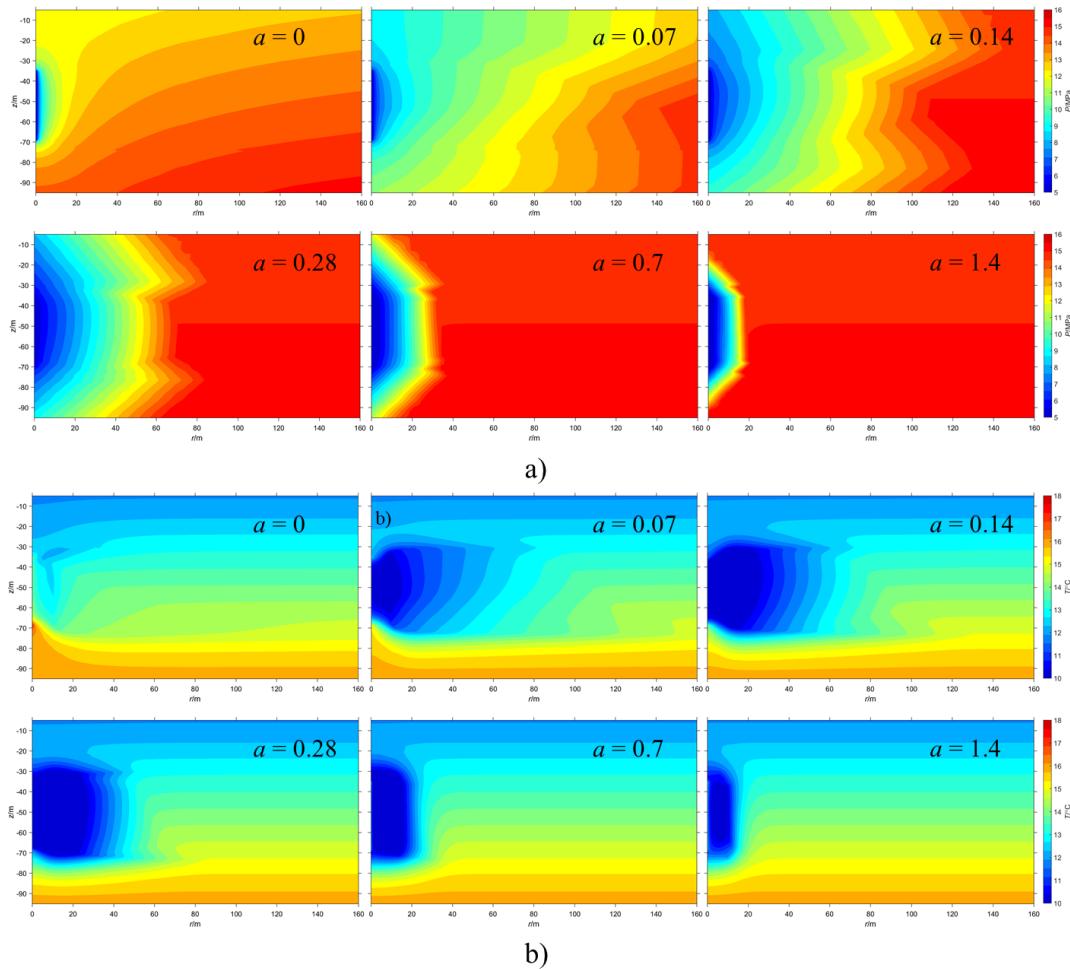
446
 447 Fig. 11 shows the gas production behavior with TPGs with a constant
 448 bottom-hole pressure (4.5 MPa). Under depressurization without TPG, it is illustrated
 449 that the gas production rate is relatively low and declines over time. The output result
 450 is optimal when $a = 0.28$, and the TPG $\lambda = 1.0166$ MPa/m, which is twice the
 451 experimental value. In this case, the gas production rate drops to 2129 sm³/d on the
 452 73rd day, then it gradually increases and stabilizes at 2700 sm³/d about 800 days later.
 453 When the value of $a = 0.7$ and $\lambda = 2.5415$ MPa/m is used, the gas production rate (772
 454 sm³/d) is close to that of the case without TPG (664 sm³/d) at the end of the
 455 simulation. However, when $a = 0.7$, the initial gas production rate decreased slightly,
 456 and the production in the first 1000 days remained above 1000 sm³/d. A larger a in
 457 case 6 is expected to result in the obstacle for pressure propagation, and, consequently
 458 in insufficient hydrate dissociation. Consistent with expectations, the corresponding
 459 gas production has slipped, as a review of the simulation outputs indicates.



460
461 **Fig. 11 Comparison of gas production rate at different starting pressures**
462

463 The distribution of pressure, temperature on the 1800th day of production with
464 different TGP's are depicted in Fig. 12.

465 As is a common occurrence in the case without TGP that the pressure drop
466 spreads rapidly across the formation as time advances. From Fig. 12a, it can be seen
467 that the pressure front can reach 150 m away. In the cases considering TGP by Eq. (14)
468 with different a , the pressure distribution presents various patterns. It is mainly
469 reflected in two aspects. Firstly, in the vertical direction, pressure advance front
470 behaves as a radial extension. Secondly, in the lateral direction, the TGP acts like that
471 it could make the pressure of far-field formations stay “frozen” and make the pressure
472 drop only occur near the well. Comparing different graphs in Fig. 12a, an important
473 observation is that the depressurization is faster near the well with a relatively small a ,
474 so hydrates can dissociate more efficiently and cooling is more pronounced at this
475 location. However, when the TGP gradually increases, the lateral extension of the
476 pressure drop is gradually blocked, making the hydrate far away from well hardly
477 reach its equilibrium pressure. Especially when $a = 1.4$ (6th graph in Fig. 12b), hydrate
478 dissociating zone is limited within 20 m after 1800 days of production, with the
479 shrinkage of progressive cooling zone caused by continuing dissociation.



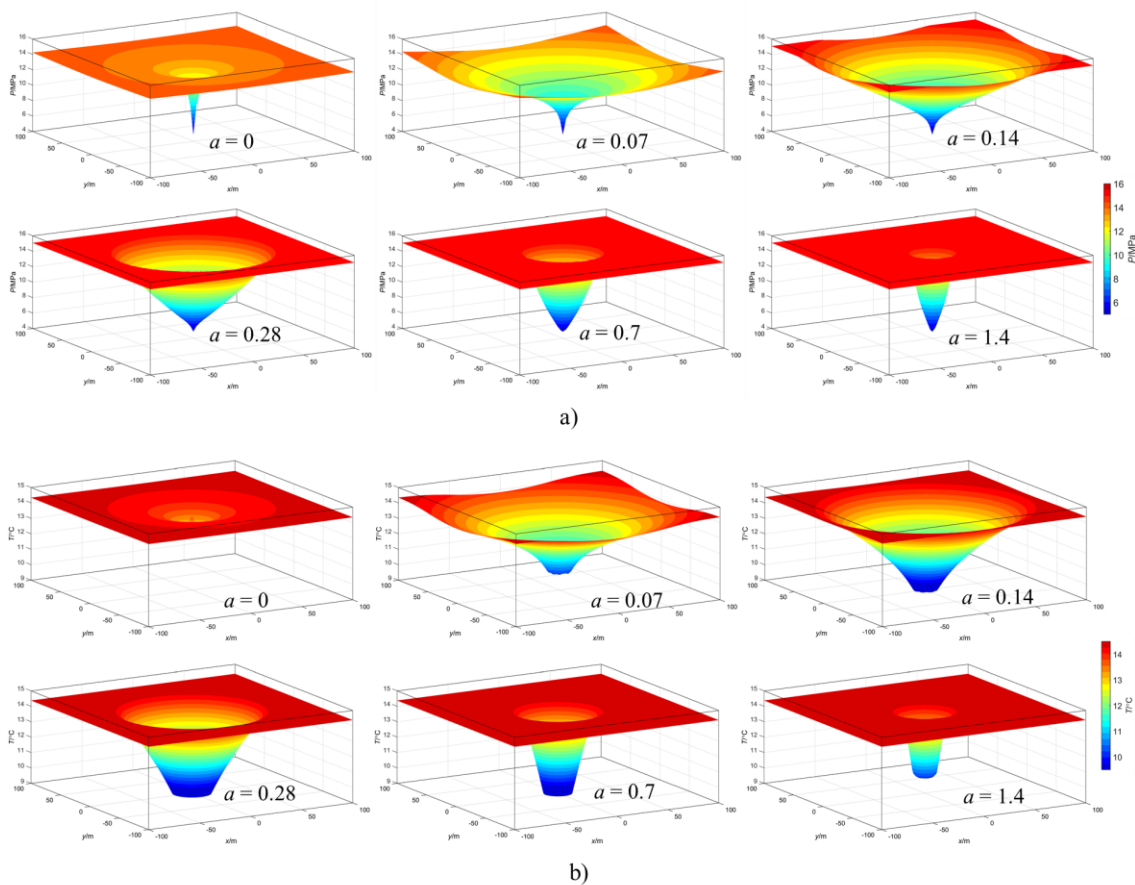
480

481 **Fig. 12. (a) Pressure distribution and (b) temperature distribution with different TPGs on**
 482 **the 1800th day.**

483

484 Fig. 13 show the pressure and temperature drop funnel (100 m away from the
 485 well, -60 m in the model) with different TPGs after 1800 days of production. It can
 486 be demonstrated from Fig. 13a that the radius of pressure drop funnel reduces with the
 487 increase of TPG. For the temperature drop funnel exhibited in Fig. 13b, the greater the
 488 TPG, the smaller the radius of funnel. However, by comparison of different graphs in
 489 Fig. 13b, the depth of temperature drops funnel augments firstly, then decreases with
 490 the rise of TPG. The minimum temperature in the profile is about 10 °C and occurs in
 491 the vicinity of the production well. It is a favorable situation that the temperature in
 492 the reservoir declines by 5 °C (from its original 15 °C). That indicates that, under the
 493 conditions in this study, the temperature drops are mild and the risk of evolution of ice
 494 is minimal. For the presence of ice in the hydrate reservoirs may reduce the effective
 495 permeability or even block the flow path of the fluid phases to the proximity of the
 496 well, which will have a significant influence on gas production, especially for the gas

497 production in the short term [50, 51].



498

499 **Fig. 13. (a) The pressure drop funnel and (b) temperature drop funnel on the 1800th day**
500 **with different TPGs.**

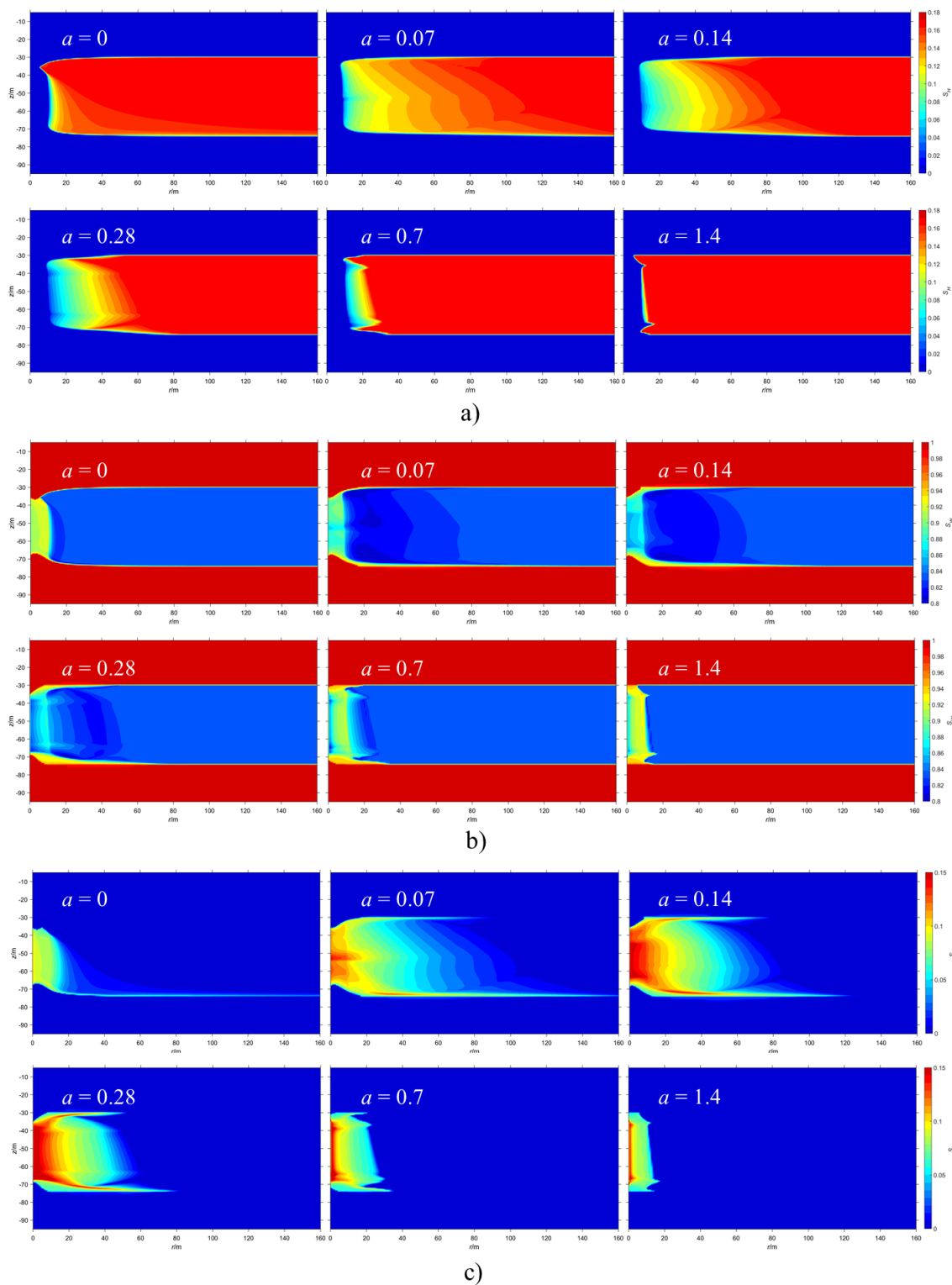
501

502 Fig. 14 show the distribution of hydrate saturation, liquid phase saturation and
503 gaseous phase saturation after 1800 days of production with different TPGs. The front
504 of hydrate dissociated zone without TPG is 15 m away from the well. When
505 considering the TPG, the front of active dissociation regions in reservoir is greatly
506 extended. That means the range of three-phase zone, where gas, water and hydrate
507 coexist, expands remarkably. In 2nd graph of Fig. 14a, the hydrate active dissociation
508 area has expanded to 110 m away from the well. However, with the TPG gradually
509 increasing, the width of the three-phase zone is gradually reduced. There is almost no
510 three-phase zone, as the 6th graph in Fig. 13a confirms.

511

512 As can be seen in Fig. 14b, there is a highly water-saturated region around the
513 production well in the case without TPG. When the TPG is relatively small (Fig. 14b),
514 the highly water-saturated region around the well disappears and is replaced by the
gas-rich area (Fig. 14c). When the TPG continues to increase, the highly

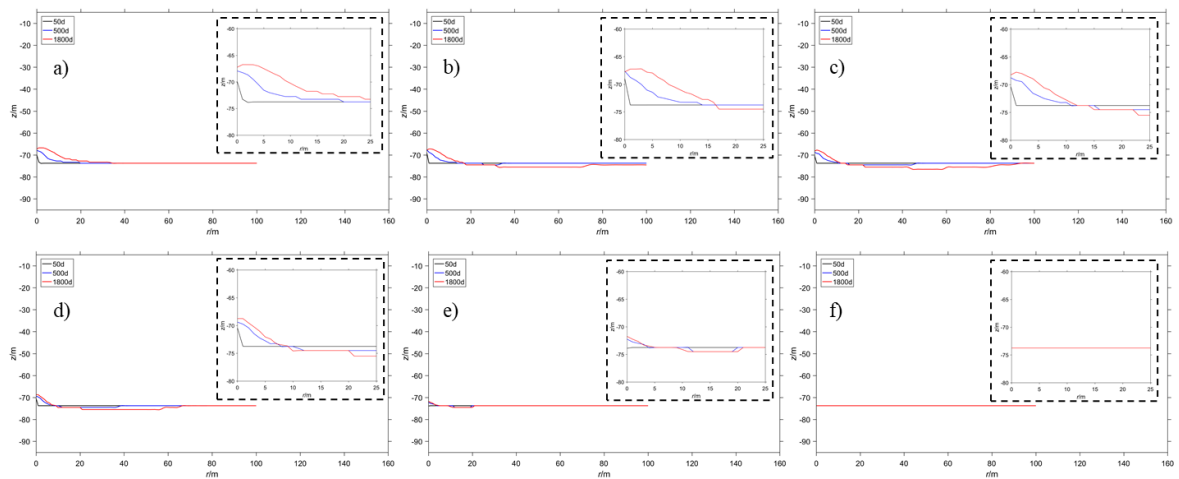
515 water-saturated region reappears, but there is still a gas-rich area next to the well.



516
517 **Fig. 14. (a) Hydrate saturation distribution, (b) liquid phase saturation distribution and (c)**
518 **gaseous phase saturation with different TPGs on the 1800th day.**

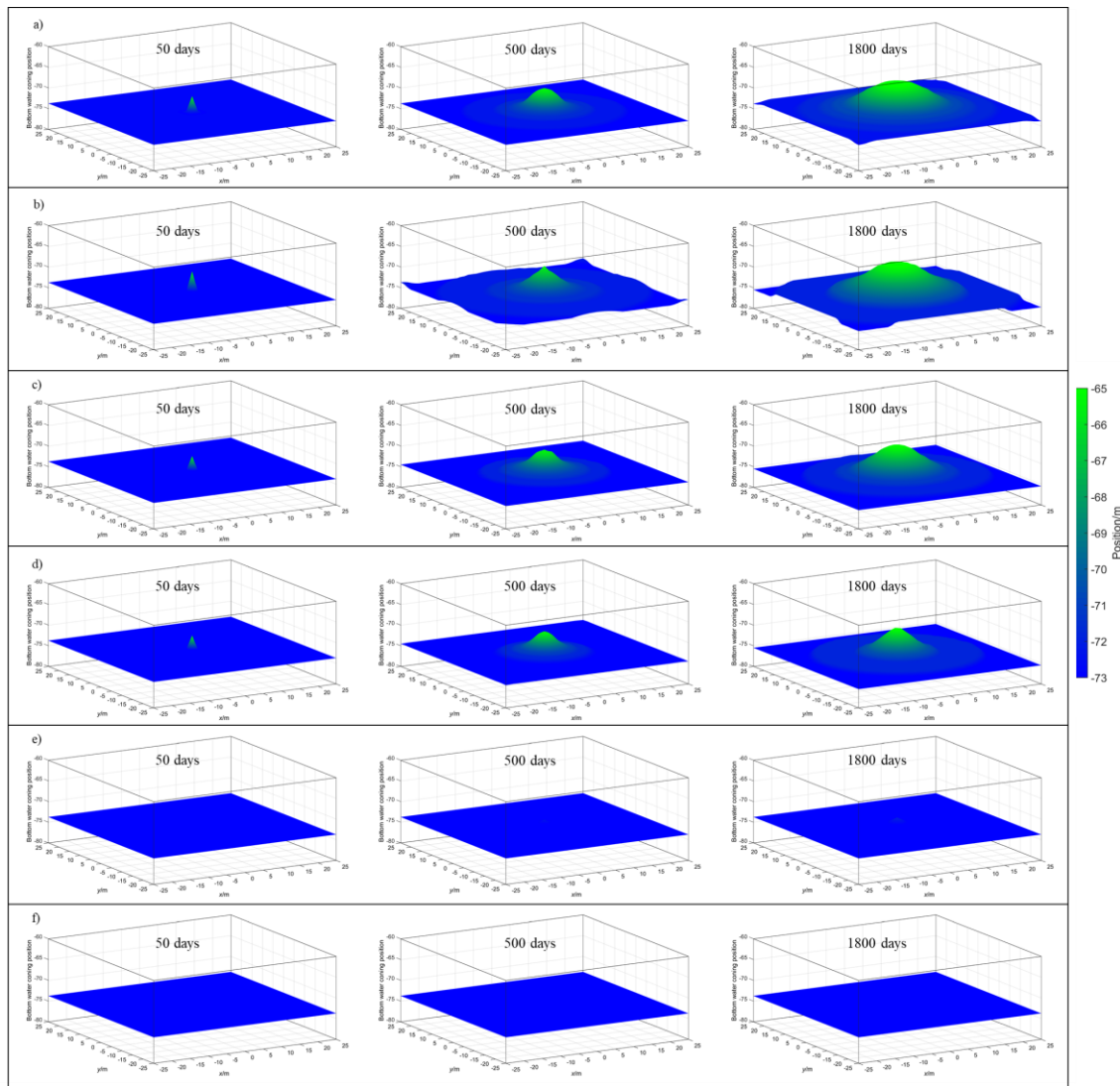
519 Because the production well did not completely penetrate the hydrate layer, the
520 bottom water invasion appears in the results [52]. Fig. 15 shows the comparison of the

521 bottom water evolution under different TPGs. Fig. 16 shows the 3D patterns of
 522 bottom water evolution. The aquifer invading into hydrate layer is obvious in the case
 523 without TPG. On the 1800th day, the height of aquifer reaches 7 m and the radius of
 524 that exceeds 25 m. In Fig. 15b, the height of aquifer does not decrease significantly,
 525 but the radius reduces to about 16 m. With the TPG increases, the height of the
 526 invading aquifer gradually decreases and the radius also shrinks rapidly. The height in
 527 Fig. 15e has dropped to lower than 2 m and the radius has shrunk to smaller than 5 m.
 528 The invasion of bottom water has even disappeared, as Fig. 16f shows.



529

530 **Fig. 15. Comparison of bottom water invasion evolution under different TPGs, a) $a = 0$, b) a**
 531 **$= 0.07$, c) $a = 0.14$, d) $a = 0.28$, e) $a = 0.7$, f) $a = 1.4$.**
 532



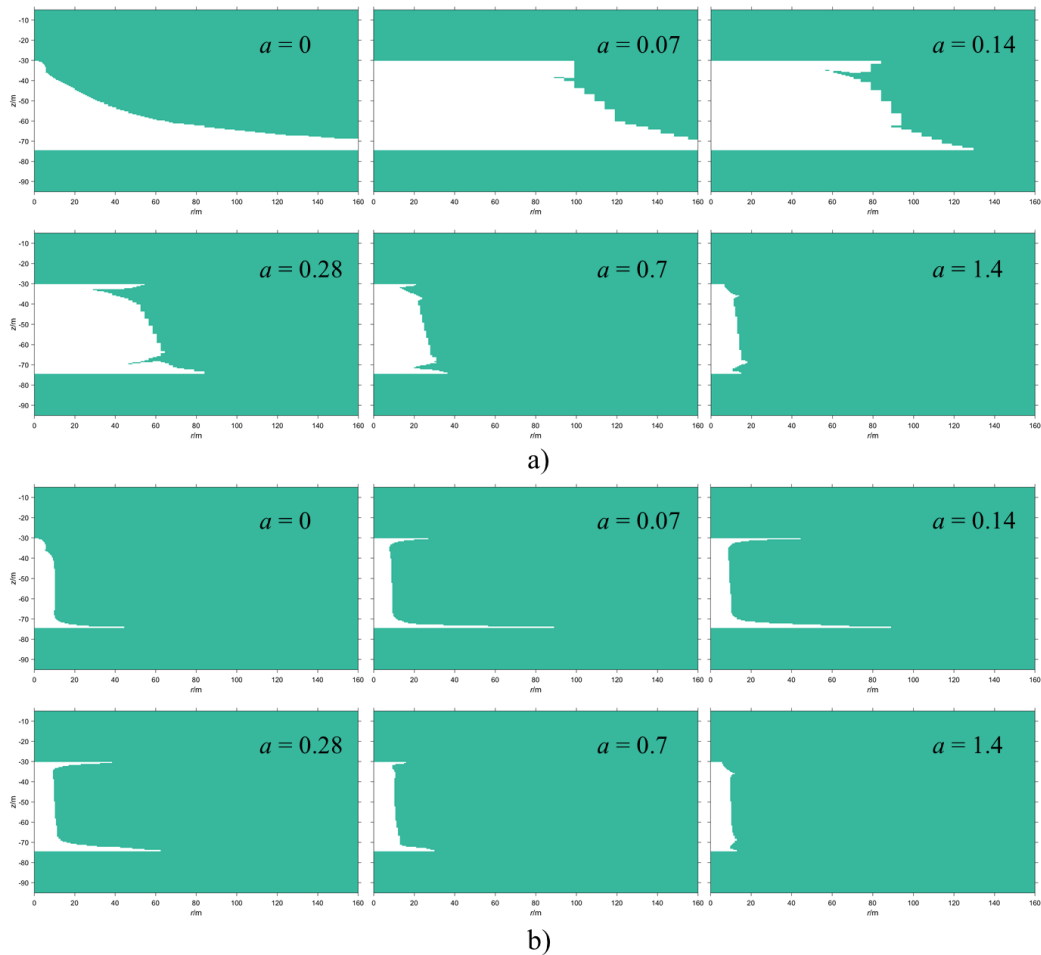
533

534 **Fig. 16. Comparison of 3D patterns of bottom water invasion evolution under different TPGs,**
 535 **a) $a = 0$, b) $a = 0.07$, c) $a = 0.14$, d) $a = 0.28$, e) $a = 0.7$, f) $a = 1.4$.**

536

537 Fig. 17a shows the advance front of the active dissociation region after 1800
 538 days of production under different TPGs. The blank part is the area where the hydrate
 539 saturation is smaller than initial value. The larger the TPG, the smaller the blank part.
 540 Review of the simulation results from Fig. 17a indicates that the TPG plays a vital
 541 role in improving the recovery of hydrate in the upper part of the hydrate layer. Fig.
 542 17b shows the advance front of the hydrate dissociated region after 1800 days of
 543 production under different TPGs. There is no hydrate in the blank area. Although the
 544 pressure, temperature, and saturation distribution of the reservoir are quite different
 545 under different TPGs, the width of hydrate dissociated region is roughly the same.
 546 The only difference lies at the interface of the hydrate layer and cover layer. As the
 547 TPG accelerates the pressure drop in the depressurization area, the hydrate at the

548 interface can completely dissociate in response to the sufficient heat supply from the
 549 upper and lower layers. As the active dissociation region is restricted by the heat
 550 supply, large area of hydrate cannot dissociation completely. Combining Fig. 17a and
 551 17b, it can be seen that as the value of a increases, the distance between the front of
 552 dissociated area and that of active dissociation region becomes smaller and smaller.



553
 554 **Fig. 17 The front of (a) active dissociation region and (b) dissociated area after 1800 days**
 555 **under different TPGs, a) $a = 0$, b) $a = 0.07$, c) $a = 0.14$, d) $a = 0.28$, e) $a = 0.7$, f) $a = 1.4$.**
 556

557 **6. Conclusions**

558 The following conclusions can be drawn from this study:

- 559 (1) The threshold pressure gradient in hydrate reservoir with argillaceous silt
 560 sediments has been quantified from the flow experiment. The TPG and the
 561 permeability-viscosity ratio agree with the power function.

562 (2) The conservation equations have been modified and the TPG module was
563 added to TOUGH+HYDRATE, where the above result of target hydrate-bearing
564 sediment in China South Sea was used. Running this function would increase time
565 consumed.

566 (3) The TPG plays a critical role in gas production from hydrate-bearing
567 sediments with unconsolidated argillaceous siltstones. When the TPG of experimental
568 value is applied, targeting at hydrate reservoir at the site SH2, production capacity has
569 been enhanced unexpectedly. It is responsible for progressive gas production rate,
570 depressed water production and relatively great gas-water ratio. The pressure advance
571 front behaves as a radial extension in the vertical direction, within which the pressure
572 drop near well is stronger. Compared with that without TPG, the front of hydrate
573 completely dissociated area has not moved forward. However, the active hydrate
574 dissociating area (gas, water and hydrate coexist) has notably extended, which results
575 in progressive cooling zone caused by continuing dissociation. Meanwhile, the TPG
576 also contributes to a slightly water-saturated zone around the well, instead of a highly
577 water-saturated zone that may cause water blockage.

578 (4) From the sensitivity analysis, the TPG with a small value can effectively
579 expand the gas output, which can expand the three-phase coexisting area mentioned
580 above. However, when the TPG further increases, propagation of pressure would be
581 restrained seriously and the pressure drop only occurs near the well. It certainly
582 accelerates the depressurization near the well, but the pressure and hydrate of far-field
583 formations stay “frozen”. That means, with increasing gradually, the TPG converts
584 from an assistance to an obstacle.

585 (5) When the production well does not completely drill out the hydrate layer, the
586 bottom water invasion would appear. The TPG can restrict the aquifer invading into
587 hydrate layer.

588 In this work, we first formatted the TPG model based on room temperature core
589 flooding test and then investigated the influence of TPG on multi-phase seepage in

590 argillaceous siltstone reservoir after hydrate decomposition using numerical
591 simulation. One natural extension of this work is to formulate more sophisticated TPG
592 model under much complicated in-situ conditions, such as various temperature,
593 pressure and hydrate saturation.

594

595

596 **CRedit authorship contribution statement**

597 **Cheng Lu:** Conceptualization, Methodology, Resources, Funding acquisition,
598 Supervision. **Xuwen Qin:** Conceptualization, Investigation. **Chao Ma:** Investigation,
599 Software, Writing - original draft, Validation, Formal analysis. **Lu Yu:** Investigation,
600 Software, Data curation, Writing - original draft & review, Validation, Formal analysis.
601 **Lantao Geng:** Methodology, Validation. **Keni Zhang:** Software, Writing – review.
602 **Hang Bian:** Project administration.

603

604 **Declaration of Competing Interest**

605 The authors declare that they have no known competing financial interests or
606 personal relationships that could have appeared to influence the work reported in this
607 paper.

608

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616

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